

11. Chilled iron has
 no graphite
 very high percentage of graphite
 low percentage of graphite
 graphite as its basic constituent of composition
12. Hypereutectoid steels have structures of
 pearlite alone
 phases of cementite and pearlite
 phases of ferrite and pearlite
 phases of cementite and ferrite
13. Which of the following constituents of steels is softest and least strong
 austenite pearlite ferrite cementite
14. Annealing of steel is done to impart the following property in steel
 ductility toughness hardness none
15. Mild steel can be converted into high carbon steel by using which of the following process?
 Annealing Normalizing Case hardening Quenching
16. Which one of the following is not correct
 martensite has a BCC structure
 austenite has FCC structure
 martensite is solid solution of carbon in BCC iron
 martensite which is formed during quenching is too brittle
17. A peritectic reaction is defined as
 two solids reacting to form liquids
 two solids reacting not to form a liquid
 a liquid and solid reacting to form another solid
 two solids reacting to form another solid
18. Pick up wrong property of austenite
 softness ductility magnetism malleability
19. White cast iron contains carbon in the form of
 free carbon graphite cementite ferrite
20. Following is the unique property of polymer material
 elasticity viscoelasticity plasticity none

KATHMANDU UNIVERSITY
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Level: B.E.
Year : II
Time: 2 hrs. 30 mins.

Course : MEEG 213
Semester: I
F. M. : 40

SECTION "B"
[40 marks]

Attempt ALL questions. Assume suitable data if necessary.

Q.N.1

- Define composition, structure, synthesis and processing. [2]
- Explain why single crystal silicon is currently the material of choice for microelectronics applications. [2]

Q.N.2

- Define the terms lattice, unit cell, basis and crystal structure. [2]
- Explain the difference between an amorphous and a crystalline material. [2]

Q.N.3

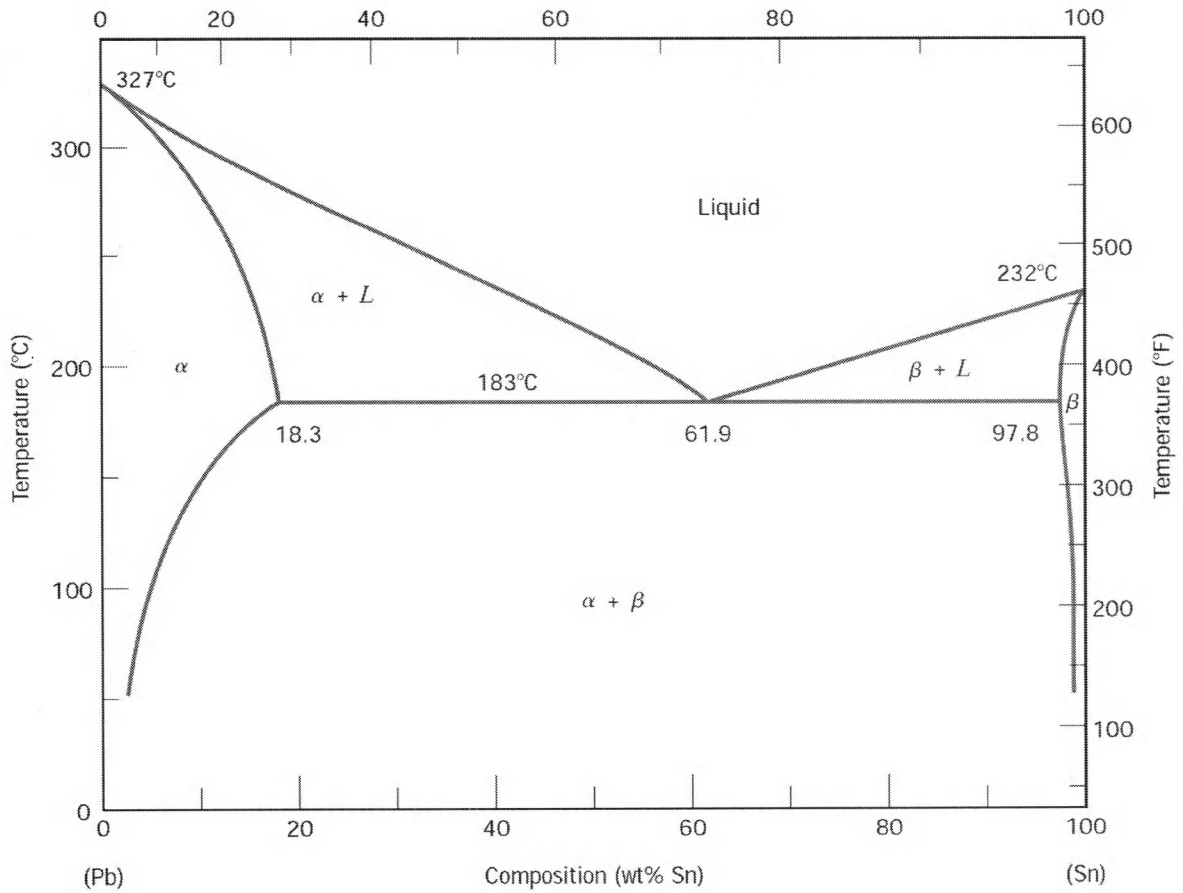
- The yield strength of mild steel with an average grain size of 0.05 mm is 138 MPa. The yield stress of the same steel with a grain size of 0.007 mm is 276 Mpa. What will be the average grain size of the same steel with a yield stress of 207 MPa? Assume the Hall-Petch equation is valid and that changes in the observed yield stress are due to changes in grain size.
The Hall-Petch equation is $\sigma_y = \sigma_0 + Kd^{-1/2}$ [2]
- Given two examples of imperfections in atomic arrangements that lead to a desirable property. [1]

Q.N.4

- Explain tensile test in detail. Discuss the properties that are obtained from tensile test with the help of stress-strain curve for ductile material. [4]
- Define impact toughness? Discuss how impact test is useful in determining ductile to brittle transition temperature. [2]
- Describe rotating cantilever beam test for fatigue failure. Explain the terms fatigue limit, fatigue strength and endurance limit with the help of S-N curve. [2]

Q.N.5

- Explain the terms superheat, recalescence, thermal arrest, total and local solidification time with the help of cooling curve for pure metal with no inoculation. [4]
- A 5-cm-cube solidifies in 4.6 min. Assume $n = 2$. Calculate mold constant in Chvorinov's rule and solidification time for a 12.5 cm x 12.5 cm x 15 cm bar cast under same conditions. [2]
- Referring to the phase diagram given, identify phases and their composition for 40 wt% Sn-60 wt% Pb alloy at 150 °C. Calculate the relative amount of each phase present in terms of mass fraction. [1+2]



Q.N.6

- Draw a self-explanatory iron-iron carbide phase diagram showing all the important phases and reactions. [3]
- Discuss the difference between full annealing and normalizing. [2]
- Draw and describe TTT diagram for steel of eutectoid composition. Explain the difference between TTT and CCT diagram. [3+1]

Q.N.7

- Write down two most important properties of ceramics. [1]
- Differentiate between polyethylene and polyvinyl chloride in terms of structure, properties and applications. [2]