

02 MAY 2023

Level : B.E.
Year : II
Time : 2 hrs. 30 mins.

Course : MEEG 213
Semester : I
F.M. : 40

SECTION "B"
[4Q. × 10 = 40 marks]

Attempt *ALL* questions.

1.
 - a. How do material properties depend upon bonding forces and energies? Explain with the help of suitable graphs. [1 + 1]
 - b. Determine the Miller indices for the planes shown in Figure 1. [3]

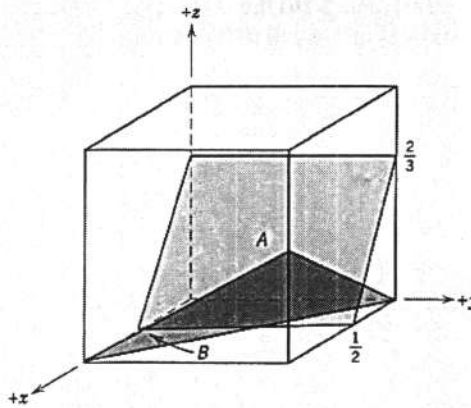


Figure 1

- c. Copper has theoretical density of 8.89 /cm^3 . It has an FCC crystal structure and an atomic weight of 63.5 g/mol . Find its unit cell edge length. [2]
 - d. Write three differences between metallic and ceramic materials. Provide relevant examples. [3]
2.
 - a. Design a cylindrical riser, with a height equal to thrice its diameter, that will compensate for shrinkage in a $5 \text{ cm} \times 10 \text{ cm} \times 20 \text{ cm}$ casting so that the riser takes 30% longer to solidify than the casting. [3]

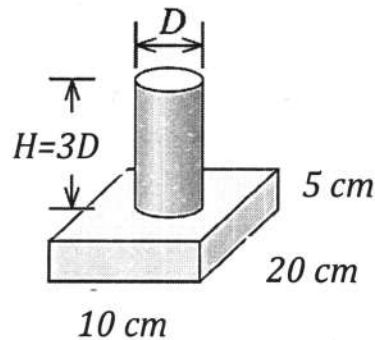


Figure 2

- b. Determine: [2 + 2]
- i. The fraction of total ferrite and cementite phases for a 99.65 wt. % Fe-0.35 wt. % C alloy at a temperature just below the eutectoid.
 - ii. Number, types and the phase amounts for 2 wt. % C at 1300°C.

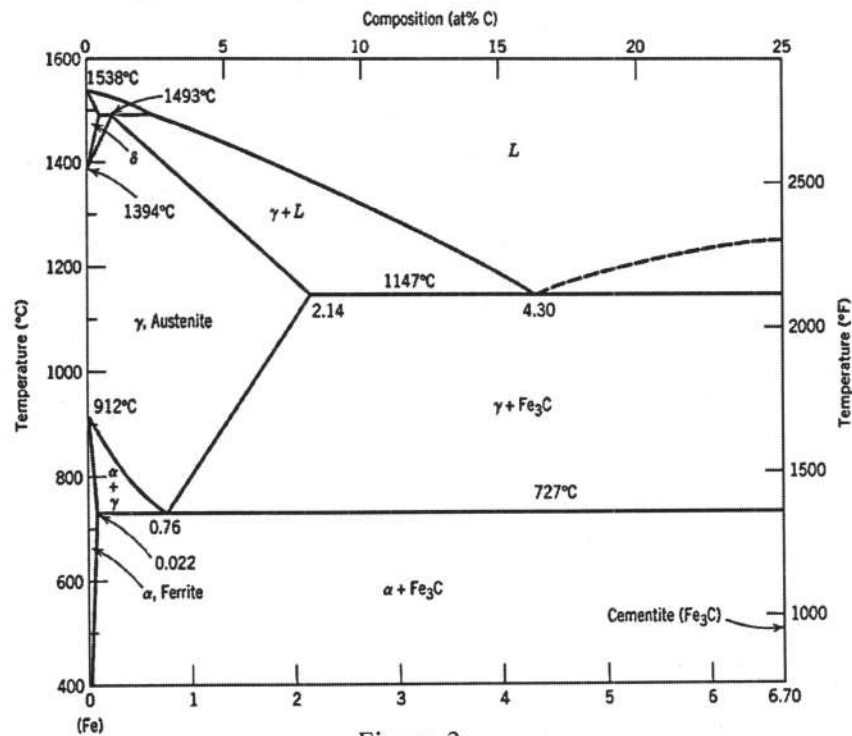


Figure 3

- c. Draw TTT diagram with important demarcations. Explain how it is formed. [2 + 1]
3.
 - a. Explain three-common types of line defects. Draw relevant figures. [2 + 1]
 - b. The metal rubidium has a BCC crystal structure. If the angle of diffraction for the (321) set of planes occurs at 27 degree (first order reflection) when monochromatic x-radiation having a wavelength of 0.071 nm is used, compute the interplanar spacing for this set of planes and the atomic radius for a rubidium atom. [3]
 - c. Apply Gibbs phase rule to find degrees of freedom for different phases and phase transformation for unary and binary systems. Interpret the results. [1 + 1]
 - d. Explain the process of twinning imperfection. Draw relevant figures. [1 + 1]
 4.
 - a. Explain 3 stages in fatigue occurrence in a material. Draw relevant diagrams. [1 + 1]
 - b. Give at least 2 precise reasons for the following: [1 + 1]
 - i. Aluminum-copper alloys are commonly used in casting industries.
 - ii. Tin is used for food packaging.
 - c. What is ductile to brittle transition? Explain with a suitable example. [1 + 1]
 - d. How are inorganic glasses formed? Explain with suitable graph. [1 + 1]
 - e. How is fracture toughness related to crack propagation? [2]