

KATHMANDU UNIVERSITY
End Semester Examination [C]
December, 2024

Level : B.Tech.
Year : II
Time : 2 hrs. 30mins.

Course : BIOT 209
Semester : I
F. M. : 55

SECTION "C"

[3Q. × 8 = 24 marks]

Attempt *ANY THREE* questions.

1. Production of single-cell protein from hexadecane is described by the following reaction equation:
$$C_{16}H_{34} + a O_2 + b NH_3 \rightarrow CH_{1.66}O_{0.27}N_{0.20} + d CO_2 + e H_2O,$$
 where $CH_{1.66}O_{0.27}N_{0.20}$ represents the biomass. If $RQ = 0.43$, determine the stoichiometric coefficient
2. Explain the general mass balance equation with an example.
3. Explain the mechanism of mixing with a well labelled diagram.
4. Explain the different types of reaction yield with respect to bioprocess technology.

SECTION "D"

[31 marks]

Attempt *ANY SIX* questions. (**Question N.O 5 is compulsory**)

5. Write short notes on (*ANY TWO*) [3+3=6]
 - a. Enthalpy
 - b. Batch process
 - c. Mixing time
6. The overall reaction for microbial conversion of glucose to L-glutamic acid is:
$$C_6H_{12}O_6 + NH_3 + 1.5O_2 \rightarrow C_5H_9NO_4 + CO_2 + 3H_2O$$

What mass of oxygen is required to produce 15 g glutamic acid? [5]
7. What is the enthalpy of 200 g formic acid at 80°C and 1 atm relative to 25°C and 1 atm? (C_p for formic acid in the temperature range = 0.524 cal g⁻¹ °C⁻¹)? [5]
8. Define batch process. Draw a diagram for typical batch growth of a bacterium. [5]
9. What are the different applications of mixing in bioprocessing? [5]
10. Explain briefly on the general energy balance equation for cell culture. [5]
11. Write the stoichiometric equation for cell growth. What is the value of respiratory coefficient (RQ) in terms of stoichiometric coefficient? [5]

